

VIRIAL EXP

$$P = \frac{RT}{V_m - b} - \frac{a}{V_m^2}$$

$$= \frac{RT}{V_m} \left[\frac{1}{1 - b/V_m} \right] - \frac{a}{V_m^2}$$

$$= \frac{RT}{V_m} \sum_{n=0}^{\infty} \left(\frac{b}{V_m} \right)^n - \frac{a}{V_m^2}$$

$$= \frac{RT}{V_m} \left[1 + \frac{b}{V_m} + O\left(\frac{1}{V_m^2}\right) \right] - \frac{a}{V_m^2}$$

$$= RT \left[\frac{1}{V_m} + \left(b - \frac{a}{RT} \right) \frac{1}{V_m^2} + O\left(\frac{1}{V_m^3}\right) \right]$$

$$B(T) = b - \frac{a}{RT}$$

COMPRESSION FACTOR

$$Z = \frac{PV_m}{RT}$$

$$P = \frac{RT}{V_m - b} - \frac{a}{V_m^2}$$

$$PV_m = \frac{RTV_m}{V_m - b} - \frac{a}{V_m}$$

$$Z = \frac{V_m}{V_m - b} - \frac{a}{RTV_m}$$

Ideal Gas

$$Z = 1$$

Now we use the reduced variables

$$P = P_c P_r$$

$$T = T_c T_r$$

$$V_m = V_c V_r$$

$$Z = \frac{P_c P_r V_c V_r}{R T_c T_r} = \left(\frac{P_c V_c}{R T_c} \right) \frac{P_r V_r}{T_r}$$

↳ $\frac{3}{8}$ FOR Van der Waals

$$Z = \frac{3}{8} Z_r \Rightarrow \frac{8}{3} Z = Z_r$$

$$Z = \frac{V_c V_r}{V_c V_r - b} - \frac{a}{R T_c V_c} \frac{1}{T_r V_r}$$

$$b = \frac{V_c}{3}$$

$$a = \frac{9}{8} R T_c V_c$$

$$Z = \frac{3 V_r}{3 V_r - 1} - \frac{9}{8} \frac{1}{T_r V_r}$$

$$Z_r = \frac{8 V_r}{3 V_r - 1} - \frac{3}{T_r V_r}$$

CHEMICAL POTENTIAL IDEAL GAS

$$\mu(T, P) = \mu^\circ(T, P^\circ) + RT \ln\left(\frac{P}{P^\circ}\right)$$

REAL GAS

$$\mu(T, P) \equiv \mu^\circ(T, P^\circ) + RT \ln\left(\frac{f}{f^\circ}\right)$$

$f \equiv$ FUGACITY

FOR ANY GAS

$$dG_m = V_m dp$$

\Downarrow

$$d\mu = V_m dp$$

$$d\mu^{\text{Ideal}} = V_m^{\text{Ideal}} dp$$

$$d\mu - d\mu^{\text{Ideal}} = (V_m - V_m^{\text{Ideal}}) dp'$$

$$\int_{P_f}^P (d\mu - d\mu^{\text{Ideal}}) = \int_{P_f}^P (V_m - V_m^{\text{Ideal}}) dp'$$

$$\mu(T, P) - \mu(T, P_f) = (\mu^{\text{Ideal}}(T, P) - \mu^{\text{Ideal}}(T, P_f))$$

$$= \int_{P_f}^P \left[V_m - \frac{RT}{P'} \right] dp'$$

$$\text{IF } P_f \rightarrow 0 \Rightarrow \mu(T, P_f) \rightarrow \mu^{\text{Ideal}}(T, P_f)$$

$$\text{OR } \frac{f}{f^0} \rightarrow \frac{P}{P^0}$$

$$f^0 = P^0$$

$$\mu(T, p) - \mu^{\text{Ideal}}(T, p) = \int_0^p \left[V_m - \frac{RT}{p'} \right] dp'$$

SAME REFERENCE CHEMICAL POTENTIAL

$$\ln\left(\frac{f}{p_0}\right) - \ln\left(\frac{p}{p_0}\right) = \int_0^p \frac{\left[\frac{p' V_m}{RT} - 1 \right]}{p'} dp'$$

$$f = p \exp \int_0^p \frac{z(p') - 1}{p'} dp'$$

$$f = p \gamma(p)$$

$$\mu(T, p) = \mu^0(T, p_0) + RT \ln(p \gamma(p))$$

$$f(P) = \phi \int_0^P \frac{z(P') - 1}{P'} dP'$$

FUGACITY COEFFICIENT

$$\frac{f}{f^0} = \frac{f(P) P}{P^0}$$

$$\lim_{P \rightarrow 0} f(P) = 1$$

$$\lim_{P \rightarrow 0} \frac{f}{f^0} = \frac{P}{P^0}$$

EQUILIBRIUM CONSTANT

$$K_f = K_p K_r$$

$$\ln K_p \frac{P_i}{P_0} \rightarrow \frac{F_i}{F_0} = \frac{\gamma_i P_i}{P_0} \rightarrow K_f$$