

$$\left(\frac{\partial H}{\partial T}\right)_p = C_p$$

$$H(T, p)$$

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$$\left(\frac{\partial H}{\partial p}\right)_T = -\frac{\mu}{\gamma T} C_p$$

$$H(T, v)$$

$$\begin{aligned} dH &= C_p dT + \left(\frac{\partial H}{\partial p}\right)_T dp \\ &= \left(\frac{\partial H}{\partial T}\right)_v dT + \left(\frac{\partial H}{\partial v}\right)_T dv \end{aligned}$$

$$dv = \left(\frac{\partial v}{\partial T}\right)_p dT + \left(\frac{\partial v}{\partial p}\right)_T dp$$

$$\left(\frac{\partial H}{\partial T}\right)_v \neq \left(\frac{\partial H}{\partial v}\right)_T \left(\frac{\partial v}{\partial T}\right)_p = C_p$$

$$\left(\frac{\partial H}{\partial v}\right)_T \left(\frac{\partial v}{\partial p}\right)_T = -C_p \frac{\mu}{\gamma T}$$

$$\left(\frac{\partial H}{\partial v}\right)_T = + \frac{C_p \mu / \gamma T}{v \kappa}$$

$$\left(\frac{\partial H}{\partial T}\right)_v = C_p \left[1 - \frac{\alpha}{\kappa} \frac{\mu}{\gamma T} \right]$$

EXTENSION TO n -DIMENSIONS

$$dF = \nabla F \cdot d\underline{r}$$

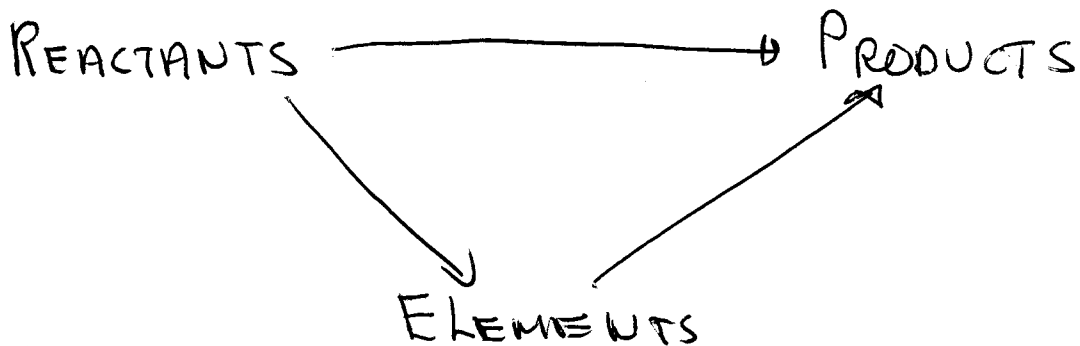
$$F(\underline{r}) = F(x, y, z, \dots)$$

$$(\nabla F)_i = \left(\frac{\partial F}{\partial x_i} \right)_{x_j, j \neq i}$$

$$dF = \sum_{i=1}^n \left(\frac{\partial F}{\partial x_i} \right)_{x_j, j \neq i} dx_i$$

$$\left(\frac{\partial^2 F}{\partial x_j \partial x_i} \right) = \left(\frac{\partial^2 F}{\partial x_i \partial x_j} \right)$$

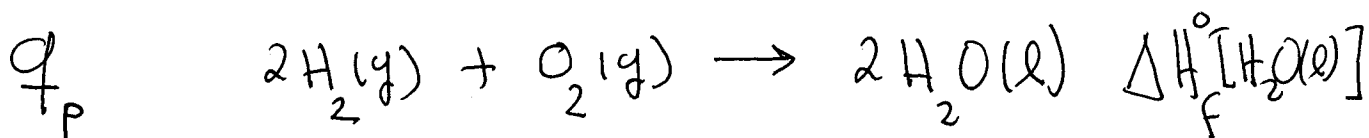
$$\nabla \times \nabla F = 0$$



ELEMENTS AT THEIR NATURAL STATE AT 1 ATM. \equiv STANDARD STATE

USUALLY WE PICK $T = 25^\circ\text{C}$ AS A REFERENCE TEMPERATURE

DEFINE FORMATION ENTHALPIES AS THE ENTHALPIES ASSOCIATED WITH RXN BETWEEN ELEMENT AT THEIR STANDARD STATE



SINCE WE CAN ONLY MEASURE DIFFERENCES IN ENTHALPY, ΔH , WE CAN SET A REFERENCE POINT.

$$\Delta H_f^\circ [\text{Elem}] = 0 \quad \text{AT } T = 25^\circ\text{C}$$

$$q_p = \Delta H_f^\circ [\text{H}_2\text{O}(l)] - 2\cancel{\Delta H_f^\circ [\text{H}_2(g)]} - \cancel{\Delta H_f^\circ [\text{O}_2(g)]}$$

THEREFORE MEASURING THE HEAT EXCHANGE IN RXN INVOLVING ELEMENTS WE CAN CONSTRUCT A TABLE OF ENTHALPIES OF FORMATION

NOW WE CAN CALCULATE ENTHALPIES OF REACTIONS AS:

$$\Delta H_{\text{rxn}}^\circ = \sum_p \nu_p \Delta H_f^\circ [\text{PRO}] - \sum_r \nu_r \Delta H_f^\circ [\text{REAC}]$$